CONTACT INTERACTION IN ZrB₂ - SiC - Cr₃C₂ CERAMIC SYSTEM

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Ultra-hightemperature ceramics on ZrB2 and HfB₂ base is a promising material for the heat and nuclear energetics, aerospace and other branches of engineering. Previously (1, 2) it was investigated the contact interaction in [ZrB₂-Cr₃C₂] system at temperatures of 1350-1600°C, were studied phase transformations and the kinetics of compaction, was fixed the eutectic nature of structure formation. It was established the formation of the diffusion zone with thickness of 5-10 microns, consisting of layers of discontinuous grained crystallites of new phases - solid solutions based on zirconium carbide and chromium borides, resulting from the exchange of boron and carbon by the initial phases. These phase interactions are responsible for the activation of syntering process and diminution of sintering temperature of porefree state obtainment for about 600°C relative to the sintering temperature of the single-phase boride of zirconium. However, in the three-phase system $[ZrB_2-SiC-Cr_3C_2]$ found a decrease of sinterability. Investigation of interaction of the components in three-phase system during hot pressing is fulfilled in the present work.

Typical structure of the diffusion zone between the Cr_3C_2 (the left side of the figure) and [ZrB₂-SiC] (right side) are shown in Fig. 1.



Fig.1 Structure of the diffusion zone. Contact interaction [800b.% ZrB₂ +200b.% SiC] and Cr₃C₂ at 1380°C and holding for 15 minutes

According MRSA, as in the case of the diffusion zone in the binary system [ZrB₂-Cr₃C], through the boundary of the primary contacts in the ternary system occurs diffusion of chromium and carbon on the one hand, and boron in the opposite direction to form the above mentioned solid solutions on the carbide of zirconium base (white grains in the diffusion zone) and chromium carbon-borides (dark grains). This transformation takes place through the eutectic liquid phases formed at the initial stages of interaction. However, the essential difference between the processes in the two-and three-phase systems is that the grains of silicon carbide dissolve in the liquid phase within the diffusion zone is the source of silicon. Silicon diffuses toward chromium carbide and forms a substantially continuous layer of chromium carbon-silicide (3) and immediately adjacent to the interface of chromium carbide.

Analysis of the results leads to the assumption that slowing the rate of sintering in the ternary system $[ZrB_2-SiC-Cr_3C_2]$ compared to double $[ZrB_2-Cr_3C_2]$ is due to the formation of this additional layer of relatively refractory chromium carbon- silicide, which restricts mutual diffusion of components through the contact boundary grains.

Work was supported by the STCU (Project P 511).

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